EORSI, M., JABIONSZKY, L., MIICH, H.

"Importance of Phage in Enteric Infections in Infants." p. 220, (NEPEGESZSEGUGY, Vol. 34, no. 8, Aug. 1953, Budapest, Hungary)

SO: Monthly List of East European Accessions, LC, Vol. 3, No. 5, May 1954/Unclassified

MORSI, M.; JABLONSZKY, L.; MILCH, H.

Significance of bacteriphage in infantile enteral infections. I. Enteritis due to E. coli 0 111 and 0 55. Acta microb. hung. 1 no.1-3:1-8 1954.

1. State Institute for Public Health and Department of Pediatrics of the Municipal Istvan Hospital, Budapest; received July 2,1953.
(MSCHMERICHIA COLI, infect.

enteritis in inf., 0 111 & 0 55 strains) (ENTERITIS, bacteriol.

M. coli in inf., 0 111 & 0 55 strains)

EBRSI, M.

EXCERPTA MEDICA Sec. 4 Vol. 10/2 Microbiology Feb 57

290. ECRSI M. State Inst. of Hyg., Budapest. *Phage types of S. typhi strains isolated in Hungary and relevant investigations made from 1950 to 1954 ACTA MICROBIOL. ACAD. SCIENT. HUNG. (Budapest) 1956, 3/3 (285-298) Tables 4

The distribution of the phage types of S. typhi in Hungary is similar to that in the other European countries. On the basis of typing 6,404 strains, the occurrence of the following phage types was proved: A, B1, B2, B3, C, D1, D2, D4, D5, D6, E1, E2, F1, F2, J, N, T, and X. Some strains proved to be sensitive to D1 + D2, D1 + D2 + D4 + D5 + D6 + N, as also some to D1 + D2 + D4 + D5 + D6 + N + L2 phages. 9.1% of the strains were untypable, and 8.2% degraded. In some cases the phage type of the same carrier's strain changed with the passing years. with the passing years. Farkas - Budapest

MORSI, M.; JABIONSZEY, L.; MILCH, H.; BARSY, G.

No translation. Acta microb. bung. 4 no.2:201-215 1957.

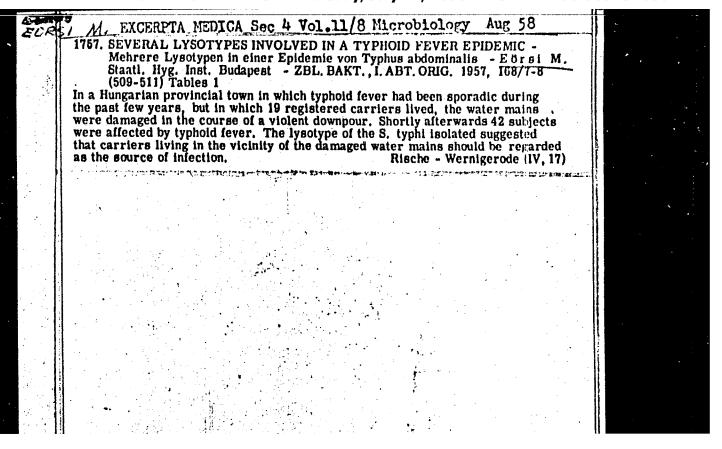
1. State Institute of Hygiene and St. Stephen's Hospital, Budapest.

(ENTERITIS, in inf. & child eticl. role of bacteriophages, feces exam. in an epidemic)

(BACTERIOPHAGE eticl. role in inf. enteritis, feces exam. in an epidemic)

"APPROVED FOR RELEASE: Thursday, July 27, 2000

CIA-RDP86-00513R00041212



MILCH, Hedda; EORSI, Maria; BOGARDI, M.

The incidence of staphylococci in hospital personnel and patients, as studied by phage-typing. Acta microb.hung. 7 no.3:285-296 160.

1. State Institute of Hygiene, Budapest, and Paul Heim Children's Hospital, Budapest.

(STAPHYLOCOCCAL INFECTIONS transm) (BACTERIOPHAGE) (HOSPITALS)

EOTTEVENYI, T.

"Electric Installations for Distance Signal Fluvioreters", p. 410 (HELMEPITESTUDOMANYI SZEMLE, Vol. 3, no. 8/9, Aug./Sept. 1953, Budapest, Hungary).

Source: Monthly List of East European Accessions, LC, Vol. 3, no. 5, May 1954/Uncl.

EOTVOS, KAROLY

Utazas a Balaton korul. (Budapest) Szepirodalmi Konyvkiado, 1957, 780 p. (Journey around Lake Balaton)

SO: Monthly Index of East European Acessions (EEAI) Vol.6, No. 11 November 1957

"APPROVED FOR RELEASE: Thursday, July 27, 2000 CIA-RDP86-00513R00041212

EFEL BRUM, G.Ye.

- 1. MIKHAILIK, A. F.: FARBER, S. G.: YEPEL'BAUM, G. Ye.
- 2. USSR (600)
- 4. Ceilings
- 7. Layer panels for ceilings. Biul. stroi. tekh. 9 no. 21, 1952.

9. Monthly List of Russian Accessions, Library of Congress, March 1953. Unclassified.

8/062/63/000/002/005/020· B144/B186

AUCHORS.

Entelis, S. C., Tiger, R. P., Nevel'skiy, E. Ya., and Epel'baum, I. V.

TITLE:

Kinetics and mechanism of the hydrolysis of carboxylic anhydrides. Communication 1. Dependence of the reaction rate on the polarity of the medium

PERIODICAL:

Akademiya nauk SSSR. Isvestiya. Otdeleniye khimicheekikh nauk, no. 2, 1963, 245 - 252

TEXT: The hydrolysis of phthalic (I) and terephthalic (II) chloro anhydride was studied spectrophotometrically at 35° C in dioxane containing 0.1 - 15.7 M/l of water. The concentration of the chloro anhydride was varied from $0.5 \cdot 10^{-5}$ to $1 \cdot 10^{-4}$ M/l. Owing excess H_2 O, the reaction seems to be zero order: $w = -do_X/dt = k_1o_X$ (2), where k_1 is the velocity constant observed and c_X is the chloro anhydride concentration during the reaction. The first order of the reaction with respect to the chloro anhydride was established from the independence of k_1 from the initial concentration. If

S/052/63/000/002/005/020 B144/B186

Kinetics and machanism of the ...

the reaction is also first order with respect to H₂Q, eq. 2 becomes w = -dc_X/ii = k₂c_Xc_{H₂Q} and k₁ = k₂c_{H₂Q}. In II, k₂ proved almost independent of the H₂Q concentration up to 0.8 M/l and then increased with increasing chapter of the two possible explanations, i.e., second-order reaction with respect to water and H₂Q effect on the dielectric constant, the first could be ruled out by plotting the curve for the rate of hydrolysis as a function of c_{H₂Q} in dioxane. To verify the second possibility, the rate of hydrolysis was studied, keeping c_{H₂Q} constant and varying the dielectric constant constant to by additing acetonitrile; k₂ increased with increasing to when the constant, k₂ also remained constant, although c_{H₂Q} increased by a factor of 3. These results for II prove that the dependence of k₂ on the H₂Q content is only due to the c_{H₂Q} effect on that the reaction is card 2/3

8/062/63/000/002/005/020 B144/B186

Kinetics and mechanism of the ...

second-order (first-order with respect to each reagent). With I, k_2 increased only in water-dioxane medium; in the ternary system, k_2 decreased with constant $c_{\rm H_2O}$ and increasing ϵ and rose slightly with constant ϵ and increasing $c_{\rm H_2O}$. For II $\log k_2 = -4.33 + 2.19(\epsilon - 1)/(2\epsilon + 1)$, and for I $\log k_2 = -3.75 + 0.91(\epsilon - 1)/(2\epsilon + 1)$. The dipole momenta calculated from these data and the Kirkwood equation were $6.95 \cdot 10^{-18}$ CGSE units for II, and $6.85 \cdot 10^{-18}$ CGSE units for I. There are 5 figures and 4 tables.

ASSOCIATION: Institut khimicheskoy fiziki Akademii nauk SSSR (Institute of Chemical Physics of the Academy of Sciences USSR)

SUBMITTED: November 15, 1962

Card 3/3

ENTELIS, S.G.; TIGER, R.P.; NEVEL'SKIY, E.Ya.; EPEL'BAUM, I.V.

Kinetics and hydrolysis mechanism of carboxyl dichlorides.

Report No.1: Reaction rate as dependent on the polarity of the medium. Izv.AN SSSR.Otd.khim.nauk no.2:245-252 F *63.

(MIRA 16:4)

1. Institut khimicheskoy fiziki AN SSSR.

(Chemical reaction, Rate of) (Anhydrides)

(Hydrolysis) (Dipole moments)

ENTELIS, S.G.; TIGER, R.P.; NEVEL'SKIY, E.Ya.; EPEL'BAUM, I.V.

Kinetics and mechanism of the hydrolysis of carboxylic acid dichlorides. Report No.2: Temperature dependence of the reaction rate, and the relation of activation energy and entropy to the polarity of the medium. Izv.AN SSSR.Otd.khim. nauk no.3:429-436 Mr '63. (MIRA 16:4) (Phthaloyl chloride) (Therephthaloyl chloride) (Hydrolysis)

TIGER, R.P.; NEVEL'SKIY, E. Ya.; EPEL'BAUM, I.V.; ENTELIS, S.G.

Kinetics and mechanism of hydrolysis of diacyl dichlorides.
Report No.3: Hydrolysis of acyl chlorides in the presence of acids and alkali. Izv. AN SSSR Ser. khim. no.11:1969-1974 N *64 (MIRA 18:1)

1. Institut khimicheskoy fiziki AN SSSR.

MAKLETSOVA, N.V.; EPEL BAUM. I.V.; ROZENBERG, B.A.; LYUDVIG, Ye.B.

Determination of molecular weight and molecular weight distribution of polytetramethylene oxide. Vysokom.soed. 7 no.1:70-73 Ja 165.

(MIRA 18:5)

1. Fiziko-khimicheskiy institut imeni Karpova, Moskva.

EPEL BAUM, Kh. I., GUTSALYUK, V. G., RAFIKOV, S. R.

"Viscosity of Paraffin-Base Petroleum at Low Temperatures," Izv. AN Kazakh. SSR, ser. khim., No 7, 1953, pp 111-117

Investigated the effect of cooling rate on dynamic viscosity for two samples of paraffin-base petroleums differing in paraffin content. Established that presence of paraffin affects structural viscosity of the petroleum. Rapid cooling of a paraffin-base petroleum produces many small crystals resulting in a large total surface which is bonded to the liquid phase, thus increasing the total volume of the solid phase, which brings about an increase in viscosity. Slow cooling produces large crystals with a smaller total surface and hence brings about a lower viscosity. (RZhKhim, No 19, 1954)

SO: Sum. No 568, 6 Jul 55

EPEL BAUM, Kh.I.; GURSALYUK, V.G.; RAFIKOV, S.R.

Influence of the residues of thermal cracking on the viscous properties of lubricating oils. Izv.AN Kazakh.SSR.Ser.khim. no.1:95-106
159. (MIRA 13:6)

(Lubrication and lubricants)

EPEL'BAUM, Kh.I.; GUTSALYUK, V.G.; RAFIKOV, S.R.

Effect of cracked stocks of the thermal cracking process on the rheological properties of paraffin oils at lower temperatures, Izv.AN Kazakh. SSR. Ser.tekh.i khim.nauk. no.1:28-35 '63. (MIRA 17:3)

GUTSALYUK, V.G.; EPEL'BAUM, Kh.I.; RAFIKOV, S.R.

Depression properties of tarry residues from petroleum refining. Izv. AN Kazakh. SSR. Ser. tekh. i khim. nauk no.2:26-33 '63. (MIRA 17:2)



EPELBAUM, M.B.

USSR/Chemical Technology - Chemical Products and Their Application. Silicates. Glass. Ceramics. Binders, I-9

Abst Journal: Referat Zhur - Khimiya, No 19, 1956, 62271

Author: Kitaygorodskiy, I. I., Keshishyan, T. N., Epelbaum, M. B.

Institution: None

Title: Effect of Heat Treatment of Mechanical Strength of Glass Fibers

Periodical: Tr. Mosk. khim.-tekhnol. in-ta, 1956, No 21, 67-73

Abstract: Different authors have found that strength of glass fibers (GF) de-

creases steadily with increasing temperature of their treatment. In this paper a study is presented of the effects of heat treatment

from GF of alkali-free and alkaline com-

USSR/Chemical Technology - Chemical Products and Their Application. Silicates. Glass. Ceramics. Binders, I-9

Abst Journal: Referat Zhur - Khimiya, No 19, 1956, 62271

Abstract: strength of GF occurs, essentially, during a short initial period of the treatment and decrease in strength is accelerated with increase in temperature. With increasing temperature of treatment strength of threads and tapes made from GF drops steadily tending asymptotically to a certain value. On increase in the duration of treatment, at a constant temperature, strength of GF decreases steadily, also tending asymptotically to a certain value. After treatment for 3-5 minutes at 7000, GF have the same tensile strength as glass in bulk. Considerations are presented to the effect that lowering of mechanical strength of GF, on heat treatment, is due to processes taking place during low-temperature crystallization of the glass and also due to increased crystallization capacity of the glass after the heat treatment.

Card 2/2

EPEL BAUN. M.R.

Installation of protective beats in furnace cooling zones. Stek. 1 ker. 14 no.3:26-27 Mr '57. (MERA 10:4) (Glass furnaces)

LUTHORS:

Keshishyan, T. H., Epel'baum, M. B.

301/ 72-58-7-4/19

TITLE:

The Structure of Glass and Its Machanical Strongth (Struktura

stekla i yego mekhanicheskaya prochnost')

PERIODICAL:

Steklo i keramika, 1958, Nr 7, pp. 12-17 (USSR)

ABSTRACT:

According to P. P. Kobeko (Ref 1) the theoretical tensile strength of silicate glass should amount to approximately 800 to 900 kilo/mm2, whereas it practically amounts to from 8 to 15 kilo/mm2 in the case of massive glass and only in the case of glass fibers of a diameter of 3 to 5μ it amounts to 400 kilo/mm2. This must be caused by the different heterogeneity of the samples which are connected with the crystallizability of the glass. This is confirmed by the work carried out by N. N. Valenkov, Ye. A. Poray-Koshita (Ref 2), O. K. Bot-vinkin (Ref 1), K. G. Kumanin (Ref 2), as well as L. I. Demkina (Ref 2) (Ref 3). The authors further investigated the influence of the glass composition with respect to its medianical strength in connection with this the diagrams are given in figures 1 and 2. Brittleness was selected amongst the mechanical properties because a method exists for its determination and since the glass samples do not require any additional heat treatment

Card 1/3

The Structure of Glass and Its Mechanical Strength 507/72-58-7-4/19

in this case. Moreover, the carrying out of these tests according to the method developed by Ya. A. Brodskiy (Ref 1) is described. The results obtained by the letermination of the brittleness are given in figure 3. As results from this, the brittleness of glass is not in a linear relation to its composition. The properties of crystallization were investigated according to the method developed by T. H. Keshishyan (Ref 1). The dependence of the crystallizability of glass and its brittleness on its composition are given in figure 4. The dependence of the brittleness on the waiting time (determined by A. Di tel) which was determined according to the method developed by Brodskiy, is graphically represented in figure 5. The existence of a certain relation between the brittleness and the crystallizability of the glass is confirmed in this way. The influence of the heat treatment on the mechanical strength of the glass fiber was investigated in the work carried out by I. I. Kitaygorodskiy, T. H. Keshishyan, M. B. Epel'baum (Ref 1). Conclusion: It was assumed that the mechanical properties of the types of glass de and in a certain way on the degree of microheterogeneity of the glass. It was tried to explain the influence of the chemical composition on the strength

Card 2/3

The Structure of Glass and Its Mechanical Strength SOV/ 72-58-7-4/19

of the glass by the change of the strength of the chemical bonds and by the crystalline force. The results obtained by the experiments carried out confirmed this. There are 5 figures and 9 references, 8 of which are Soviet.

- 1. Glass--Structural analysis 2. Glass--Mechanical properties
- 3. Glass--Test results

Card 3/3

AUTHOR:

Epel'baum, M. B.

SOV/72-58-8-3/17

TITLE:

Once More on the Usefulness of Boundary Boats in Glass Melting Furnaces (Yeshche raz o ratsional'noy ustanovke zagraditel'nykh

lodok v steklovarennoy pechi)

PERIODICAL:

Steklo i keramika, 1958, Nr 8, pp. 6 - 8 (USSR)

ABSTRACT:

I.I.Tukh maintains that the place where the boundary boats are located in the furnace was of no importance (Refs 1 and 2); the author of the present paper says, however, that this is wrong. He mentions a formula put up by A.A.Sokolov for the calculation of the consumption of flowing glass (Ref 3) and which is proved by the papers of D.B.Ginzburg and I.Peyshes. The temperatures of the mass in the cooling part drop at all depths of the basin as they approach the machines. This was also shown by the results of the investigations of furnace Nr 2 of the Gusevskoy plant imeni. Dzerzhinskiy (see figure). Then the author mentions the calculation of the viscosity as well as of the flow which he carried out on the basis of the data supplied by M.V.Okhotin (Ref 2). The glass temperature depends on the place of location of the

Card 1/3

Once More on the Usefulness of Boundary Boats in Glass Melting Furnaces

SOV/72-58-8-3/17

boat in front of it, so does the quantity of the convection current; this fact exerts an influence on the heat balance of the cooling part. The dislocation of the floating boats to the end of the cooling part is useful also from the standpoint of glass-melting technology; this view is proved by the work carried out at the Magnitogorsk glass factory. By reducing the current and the temperature of the glass the erosion of the furnace basin and of the boundary boat is decreased and the elaboration temperature in the channel is stabilized. This way the working conditions of the machines are improved which increases the quality and output glass. From the standpoint of economy the improvement of the glass quality, a better utilization of the furnace, as well as an increase of the stability of the boats must be taken into account. The editorial staff of the periodical mentions that they will discontinue the discussion on this topic until material of special investigations will be at hand. They request of the Institute of Glass to publish in the producal the resilts and conclusions of its investigations in this field. There are 1 figure and 5 references, which are Soviet.

Card 2/3

Ome More on the Usefulness of Boundary Boats in Glass Melting Furnaces

507/72-58-8-3/17

1. Glass--Melting 2. Furnaces--Performance 3. Furnaces--Equipment

Card 3/3

EPEL BAUN N.B. Calculating the temperature and maximum speed of glass orystallisation. Stek. 1 ker. 15 no.4:22-26 Ap 158. (MIRA 11:5)

> 1. Ural'skiy filial Akademii stroitel'stva i arkhitektury SSSR. (Glass manufacture)

1.13

15-(2) AUTHORS:

SOV/72-59-8-4/17 Keshishyan, T. N., Epel'baum, M. B.

TITLE:

Micro-hardness of Glass as a Function of Its Micro-heterogeneity (Zavisimost' mikrotverdosti ot mikrogeterogennosti stekla)

PERIODICAL:

Steklo i keramika, 1959, Nr 8 , pp 9-12 (USSR)

ABSTRACT:

In the experiments by A. A. Bochvar and O. S. Zhadayeva, Ye. M. Savitskiy and M. A. Tylkina, A. M. Korol'kov and E. S. Kodaner (Footnote 1) the micro-hardness method of physico-chemical analysis is used. It can be seen from the work done by A. M. Korol'kov and E. S. Kodaner, V. M. Glazov, V. N. Vigdorovich, G. A. Korol'kov, that micro-hariness is immediately connected with the phase diagram of the system (Footnote 2). In a previous paper published by the authors of the present article it was suggested that the mechanical properties of glass are conditioned by the effect of the microheterogeneity, which is due to the crystallization properties of glass and its heat treatment, upon its strength (Footnote 3). Two series of glass types were examined: first glass types of different chemical composition with the same heat treatment, and second, glass types of the same chemical composition with

Card 1/3

Micro-hardness of Glass as a Function of Its Micro- SOV/72-59-8-4/17 heterogeneity

a different heat treatment. In the investigation discussed here, 24 glass types with the same heat treatment in the system SiO2-Al2O3-CaU-MgU-Na2O were dealt with. Their microhardnesses and crystallization characteristics are shown in table 1. Their melting conditions and crystallization properties have already been discussed in the papers by T. N. Keshishyan, B. G. Varshal, Ye. A. Faynberg (Footnote 4). Yu. V. Rogozhin, Z. M. Syritskaya, B. V. Tarasov (Footnote 5) as well as N. M. Pavlushkin and G. G. Sentyurin (Fcotnote 6) used specially polished samples in the examination of the glass micro-hardness. The methods suggested by the authors consist in measuring the glass micro-hardness of fresh splinters with a grain size of 2-3 mm, whereby internal tensions in the glass are practically eliminated. A series of 24 glass samples was examined. The samples were melted and cooled under constant conditions. The second series examined was one of glass samples taken from different places of a tank furnace of the Magnitogorsk glass factory and cooled in water. The results obtained with the first series are shown in table 1, with the second

Card 2/3

Micro-hardness of Class as a Function of Its Micro- 30V/72-59-8-4/17 heterogeneity

series in table 2. Figure 1 shows the micro-hardnesses of the different glass types as functions of the maximum rate of crystal growth. In connection with these examinations the work done by P. P. Kobeko (Footnote 6) is mentioned. Figure 2 shows the micro-hardness of glass from the Magnitogorsk factory as a function of the sampling temperature. It was proved by the investigations under consideration that a change of the the investigations under consideration that a change of the degree of micro-heterogeneity of glass by different heat treatment results in a change in the mechanical properties of the glass. There are 2 rigures, 2 tables, and y references, 7 of which are Soviet.

Card 3/3

KESHISHYAN, T.N.; EPEL'BAUM, M.B.

Relation between the mechanical properties of glass and its crystallization. Trudy MKHTI no.27:150-155 *59. (MIRA 15:6) (Glass--Analysis)

EPEL'BAUM, M. B., CAND TECH SCI, "CERTAIN PROBLEMS OF THE STRENGTH OF GLASS IN CONNECTION WITH ITS TENDENCY TO CRYSTALLIZE." [MOSCOW] 1960. (MIN OF HIGHER ED USSR, MOSCOW ORDER OF LENIN CHEMICO-TECHNOL INSTITUTE. I. MENDELEYEV). (KL, 3-61, 223).

299

BRONSHTEYN, A.P.; ARKHANGEL'SKAYA, T.V.; TALISMAN, L.B.; GORBATYY, Yu.Ye.; EPEL'BAUM, M.B.

Physicochemical investigation of the thermal destruction process of some Kuznetsk Basin coals. Koks i khim. no.11:12-17 '62.

1. Chelyabinskiy metallurgicheskiy zavod (for Bronshteyn, Arkhangel'skaya). 2. Ural'skiy filial Akademii stroitel'stva i arkhitektury SSSR (for Talisman, Gorbatyy, Epel'baim).

(Kunzetsk Basin—Coal—Carbonization)

5/072/62/000/004/001/002 B105/B101

AUTHORS:

PERIODICAL:

Epel'baum, M. B., Gorbatyy, Yu. Ye.

TITLE:

Internal stresses and change of mechanical properties in

glass

Steklo i keramika, no. 4, 1962, 11 - 14

TEXT: The effect of a difference between the linear expansion coefficients of two phases was studied in glass containing spherulites. Gradually cooled glass of the Magnitogorskiy stekolinyy zavod (Magnitogorsk Glassworks) contained spherulites of 8 - 10 mm diameter. Considerable internal stresses characterized by double refraction were detected around the spherulites. From the path difference of the beams measured with a Nikitin-Berek compensator the internal stress P was calculated by using the equation $P = \Delta/N\delta$, where $\Delta_{(m_{ji})}$ is the path difference owing to double refraction, δ the thickness of the specimen in $m\mu$, and N the optical stress coefficient equated to 2.5.10-7 kg/cm2. It was found by means of a MMI-3 (PMT-3) unit that both microhardness and internal stress decrease as

Card 1/2

S/072/62/000/004/001/002 B105/B101

Internal stresses and change ...

the distance from the spherulite increases. Since chemical analysis of the spherulite and the glass in its immediate neighborhood revealed no marked difference in composition, the internal stress can be explained by the difference between the thermal expansion coefficients of glass and spherulites. Removal of the spherulite from the glass by boring resulted in a decrease of the internal stress. There are 5 figures and 4 tables.

Card 2/2

EPEL BAUM, M. B.; GORBATTY, Yu. Ye.

Method of preparation of samples for measuring the microhardness of glasses. Zav. lab. 28 no.12:1492-1494 '62. (MIRA 16:1)

1. Uraliskiy filial Akademii stroitelistva i arkhitektury SSSR.

(Glass-Testing) (Hardness)

BRONSHTEYN, A.P.; MAKAROV, G.N.; GORBATYY, Yu.Ye.; EPEL'BAUM, M.B.

Shrinkage and formation of phase stresses in coke. Koks i khim. no.8:22-27 '63. (MIRA 16:9)

1. Chelyabinskiy metallurgicheskiy zavod (for Bronshteyn).
2. Moskovskiy ordena Lenina khimiko-tekhnologicheskiy institut im.
D.I. Mendeleyeva (for Makarov). 3. Ural'skiy filial Akademii stroitel'stva i arkhitektury (for Gorbatyy, Epel'baum).

(Coke)

EPEL BACM, MIL.:

KOLOTYY, S.G.; KOTONIN, V.A.; MPEL'BAUN, M.L.; MAUNOV, P.A.

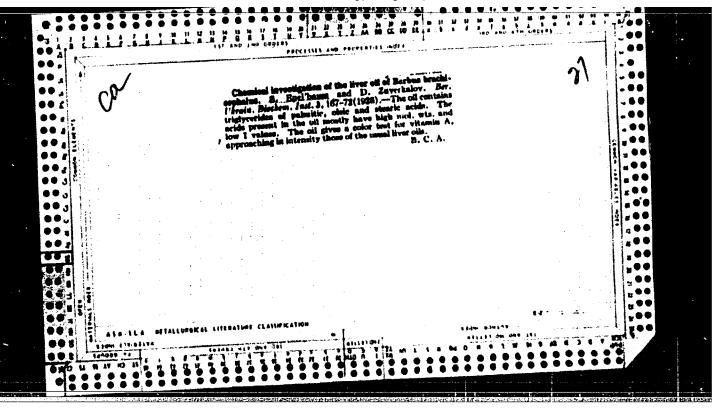
Increasing the productivity of cement mills and reducing power consumption during grinding. Prem. energ. 12 no.3:25 Mr '57.

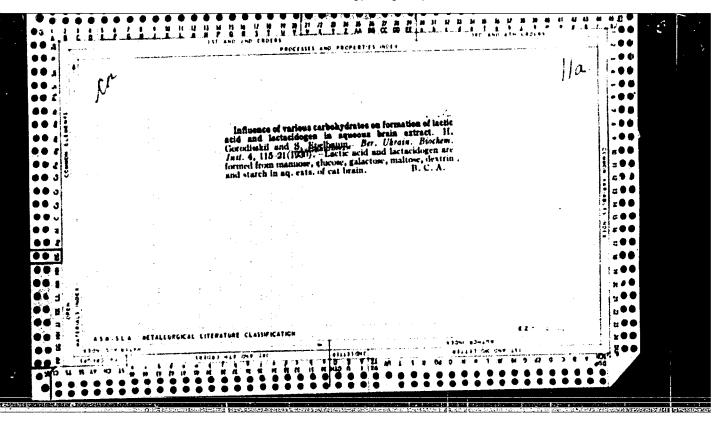
(Cement plants) (MIRA 10:4)

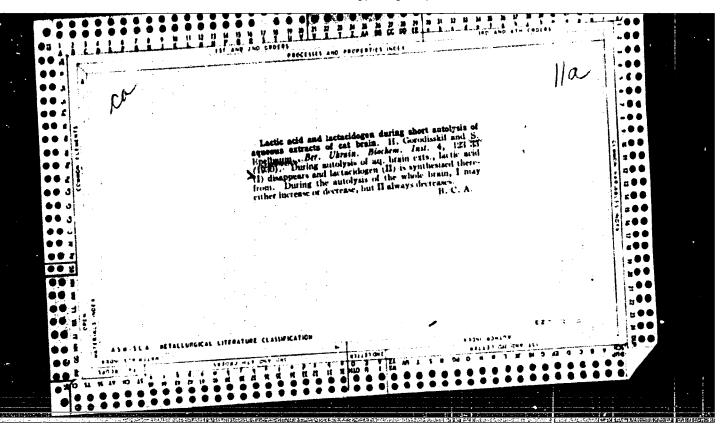
EPFEL'BAUM, R.V.

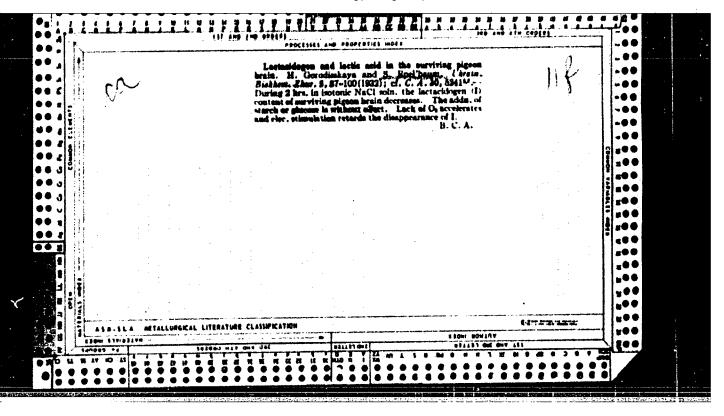
New types and designs of water heating equipment. Vodopod., vod. resh. i khimkont. na parosil. ust. no.1:62-86 *64. (MIRA 18:2)

1. Moskovskoye otdeleniye TSentral'nogo nauchno-issledovatel'skogo i proyektno-konstruktorskogo kotloturbinnogo instituta im. Polsunova.

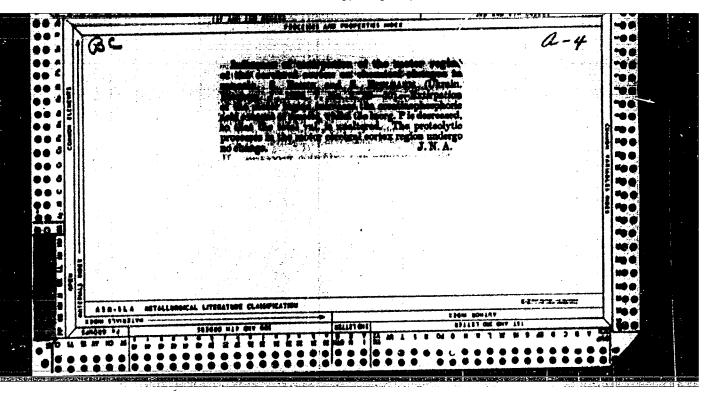


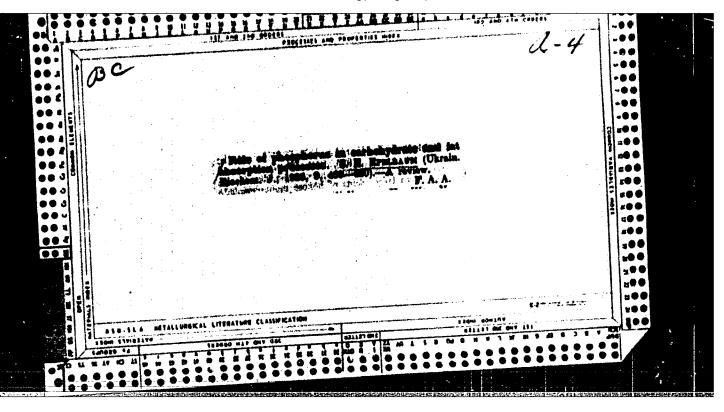


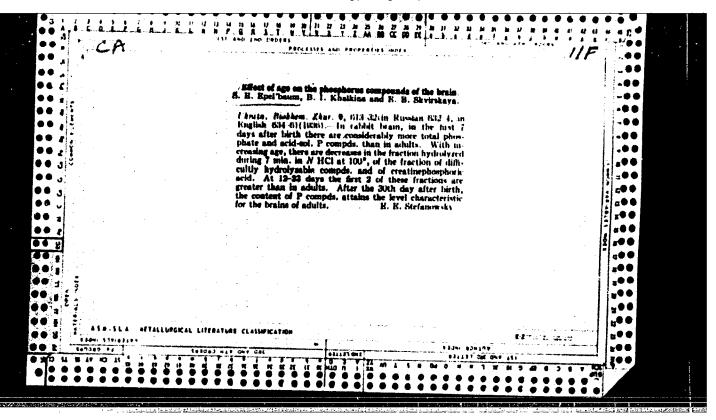


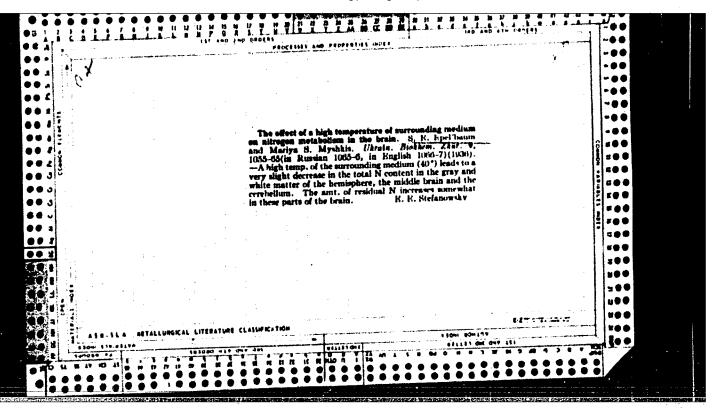


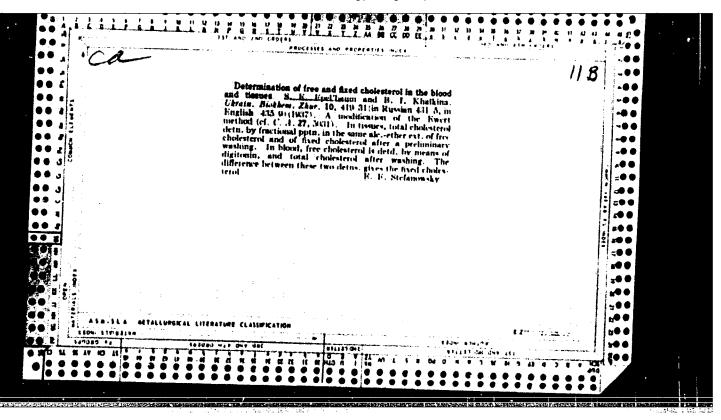
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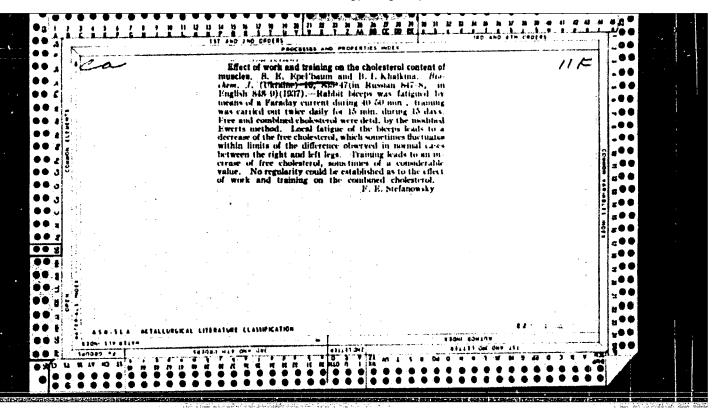






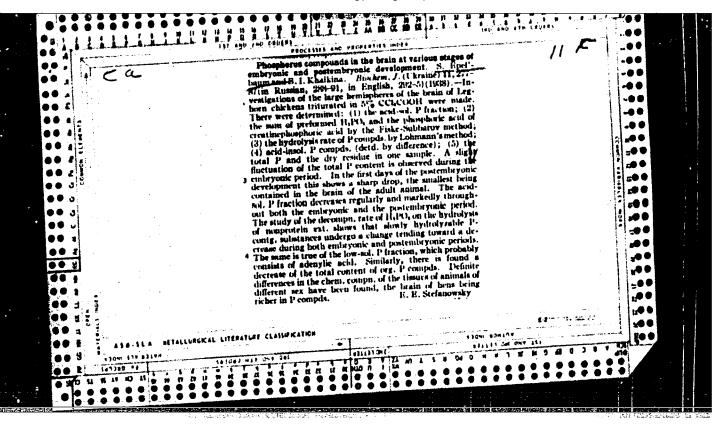


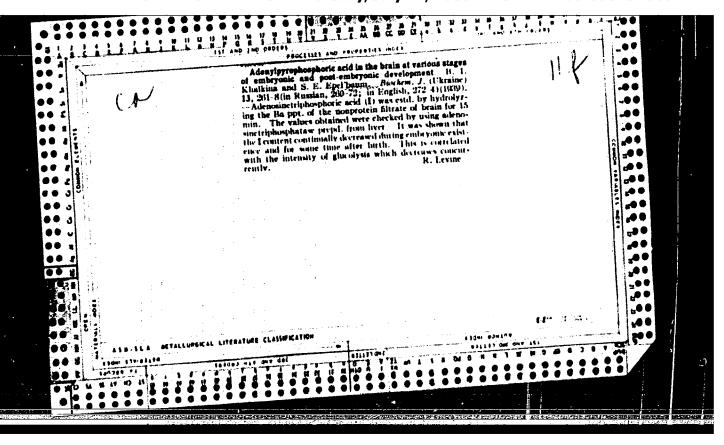


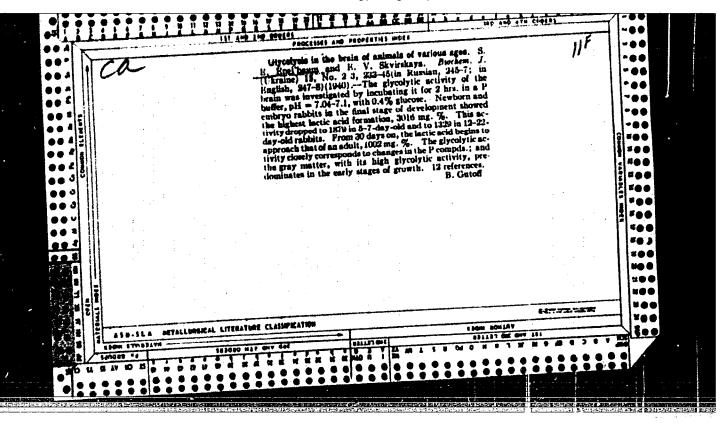


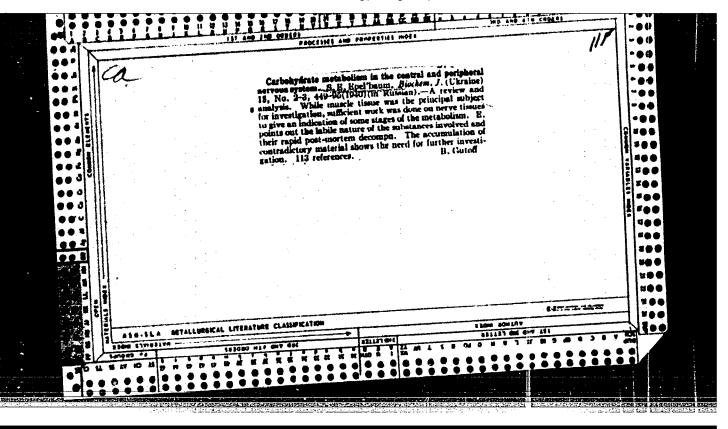
"APPROVED FOR RELEASE: Thursday, July 27, 2000

CIA-RDP86-00513R00041212









EPEL'BAUM, S. YE.

PA 3/50160

pen / Medicine - Biotin Poods Nov/Dec 48

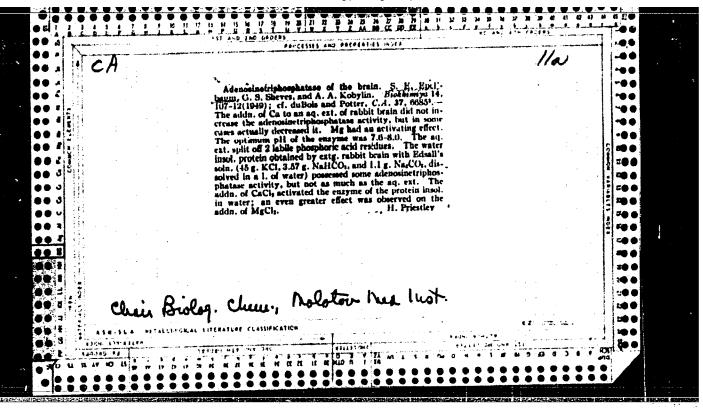
"Biotin, Its Chemical Nature and Biological Role," S. Te. Epel baum, Moscow, 17t pp

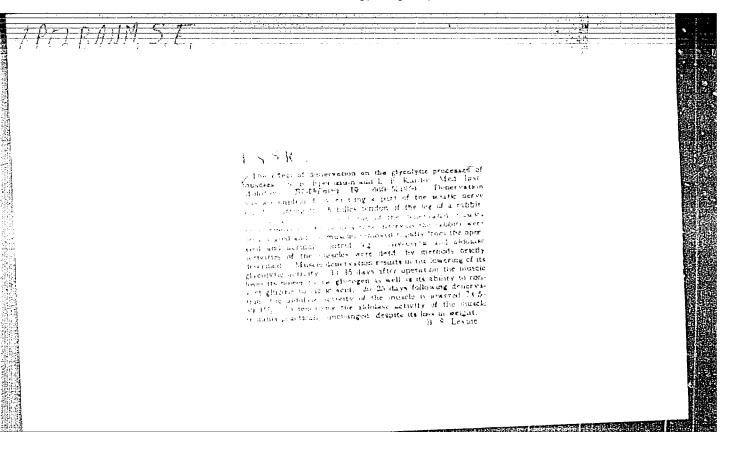
"Uspekh Sovrem Biol" Vol XXVI, No 3 (6)

Gives history of biotin. Tables show content, distribution of liver lipins, effect of various admixtures on development of diseases in chickens, and amounts of biotin in various foodstuff. Further research should reveal its biological processes, mechanism of action, and formation.

3/50160

Chair of Biological Chaus, Moloton had I wot.





"APPROVED FOR RELEASE: Thursday, July 27, 2000

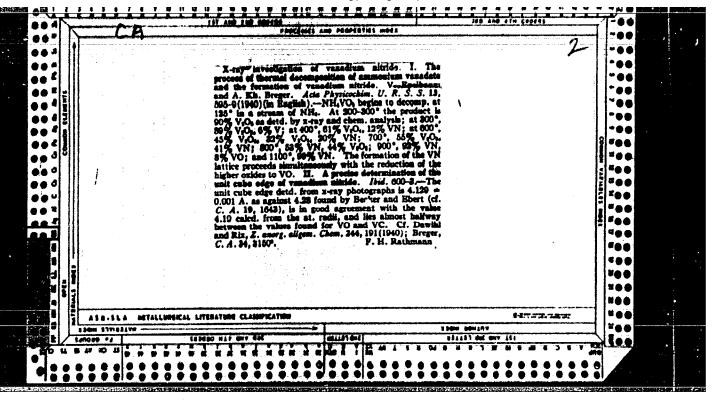
CIA-RDP86-00513R00041212

Protein inegabotism and the processes of oxidation in the denervated muscles of hypothyroid animals (1.8 however). F. E. Epc Phonin, and V. I. Kyrmino (Med. Inc. M. I. e. S. Baskeroid) TT-T-17-(18-50). The search near control of rabbuts was severed and the nerve on the order side was left intent for a control. A part of these rabbuts rear of for a min. of 30 days 6-methylthorizated at the rate of 160 mg./kg. leady wt. before and after the operation. F. Clow ling the denervation the rabbuts were kilied at metroals by decapitation and the gasticeneous and plantaris unused were examid. The degree of proteological and the rate of medicine of labeled methodics. O absorption was detal with the rate of of the Warburg app. Pyrophophatase acrieval; (I was also detal. Muscle denervation led to a more intensive inclusion of labeled methodings into the protein of the affected muscles; respiration and I activity were cultivated in denervated muscles of hypothyroid animals muscular arrophy was impeded while the protectyle activity was somewhat increased. The rate of inclusion of radicalebeled methodine remained unaffected as did the rate of O absorption. The activity of I was increased to a slight degree.

B. S. Leyne



uni arregare



EPEL'BAUM, V. A.

PA 18TEO

USSR/Chemistry - Vanadium Compounds Chemistry - Systems, Binary Jun 1946

"I-Ray Studies of Vanadium Nitride--III: The System VN - VO," V. A. Epol'baum and A. Kh. Breger, 2 pp

"Zhur Fiz Khim" Vol XX, No 6

Discussion, with accompanying graphs and tables, leading to the conclusion that the exposure time of solid solutions in the binary system VN-YO undergoes a linear change from the exposure time of pure vanadium nitride (4,129A) to the exposure time of pure vanadium oxide (-4,08A), with relation to the concentration of the components.

18780

EPELBAUM. V.

PA 52T9

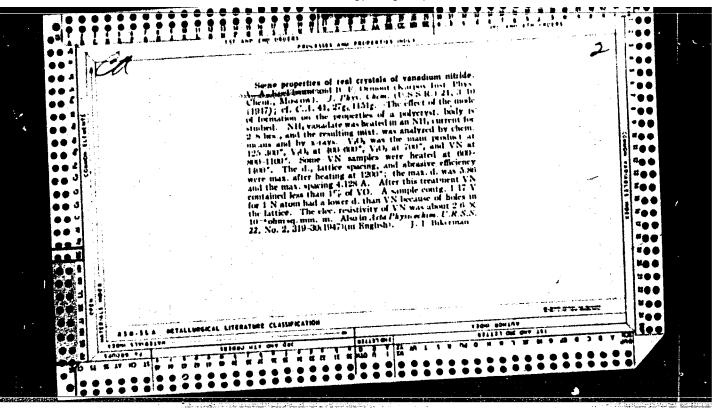
USSR/Chemistry - Vanadium Mitride Jul/Aug 1946 Chemistry - ISBey Study

"An X-Ray Examination of Vanadium Nitride. III. The System VN--VO," V. Epelbaum, A. Brager, X-Ray Iab and Iab of Splid Compounds, Karpov Inst Phys Chem, Moscow, 3 pp

"Acta Physicochimica URSS" Vol XXI, No 4

Shows unit cube edge of solid solutions VN--VO changes linearly with the concentration of the compounds from the unit cube edge of pure vanadium nitride (4.129 Å) down to that of pure vanadium oxide (~4.08 Å). Received 15 Aug 1945.

5219



EPELBAUM, V.

PA 9719

USER/Crystals - Properties Crystals - Growth Fob 1947

"Certain Properties of Real Crystals of Vanadium Nitride," V. Epelbaum, B. Ormont, 12 pp

"Acta Physicochimica" Vol XXII, No 2

Study of the reaction of the formation of real crystals of vanadium nitride and their physicomechanical properties, in relation to the conditions under which they were formed, to establish the influence of these conditions on crystalline structure and properties.

9T19

EPEL'BAUM, V. A.

PA 61T8

OGER/Chemistry - Bitrates - Detection Chemistry - Analyses - Methods

Jan 1948

"An Analysis of Vanadium Mitride," V. A. Epel'baum, B. F. Ormont, Phys Chem Inst imeni L. Ya. Karpov, 12 pp

"Zavod Labor" Vol XIV, No 1

Brief description of Dyum's, Kjeldahl's, and alkali method for determining amount of nitrogen in various compounds, particularly in nitrates.

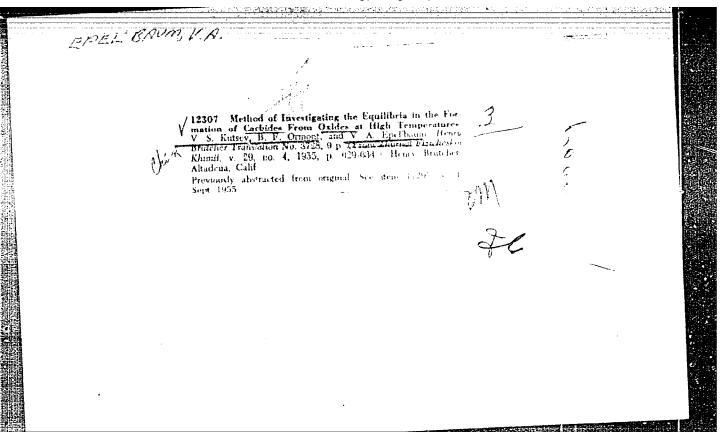
61**T**8

EPELBAUM, V.A.

Distr: 4E2c

The system boron-carbon-allicon and obtaining of "borundum." B. F. Ormont, V. A. Boel haum, and I. C. Shafran. Bor, Trudy Konf. Rhim. Bori & Ego Soedinenti 1955, 177-81 (Pub. 1988).—Abrasive properties were investigated in 5 products contg. B 12.1-45.8, C 13.3-38.2, and Si 29.8-62.3% and prepd. in a Tamman furnace by a chem. reaction between BrO., SiO., and C. The products are described in the following stoichiometric formulas: Si₁B₁C, SiB₂C, SiB₂C, SiB₃C, SiB₄C, SiB₄C, SiB₄C, as raw materials, boric acid, swaged white silica, and soot were employed. The borundum products had high abrasive properties, and their production cost appeared to be many times lower than that of carborundum, as the consumption of valuable raw materials was greatly reduced.

W. Tomaszczyk.





QUREVICH, M.A.; KUTSEV, V.S.; ORMONT, B.F.; SMIRNOVA, V.I.;

EPEL'BAUM, V.A.

Variable-composition phases in the chemistry of carbides.

Variable-composition phases in the chemistry of carbides.

(MLRA 9:11)

Zhur.neorg.khim. 1 no.7:1578 Jl '56.

Profession

64 Pro 1 Te

EPEL DAMM, YH,

"Concerning the Formation of the Phase SiB₃ in the System Silicon-Boron," by M. A. Gurevich, V. A. Epel'baum, and B. F. Ormont, <u>Zhurhal Neorganicheskoy Khimii</u>, Vol 2, No 1, Jan 57, pp 206-208

On the basis of the experimental results reported, the authors dispute the conclusion by G. V. Samsonov and V. P. Latysheva (<u>Doklady Akademii Nauk SSSR</u>, Vol 105, 1955, p 499) to the effect that only one boride phase, the composition of which corresponds to the formula B3Si, is formed in the system silicon-boron. They maintain that the phase SiB3 also exists.

[Comment: Silicon borides are of importance as semiconductor materials,

MPEL'BAUM WAREN YANOV, N.G.; GUERVICH, N.A.; ORMONT, B.F.; ZHDANOV,

Phases formed in the system chromium -- boron. Part 1: Formation of "\$\beta\$-chromium" under the influence of small additions of boron.

Zhur. neorg. khim. 2 no.8:1848-1854 Ag '57. (MIRA 11:3)

(Chromium) (Boron)

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68954 SOV/81-60-2-4306

Translation from: Referativnyy zhurnal. Khimiya, 1960, Nr 2, p 103 (USSR)

AUTHORS:

Kudintseva, G.A., Tsarev, B.M., Epel'baum, V.A.

TITLE:

The Borides of the Transition Metals and Their Electron-Emission

9.31.20

68953

10**V/81-60-2-**4305

Translation from: Referativnyy zhurnal. Khimiya, 1960, Nr 2, p 103 (USSR)

AUTHORS:

Kudintseva, G.A., Epel'baum, V.A., Tsarev, B.M.

TITLE:

The Synthesis of Hexaborides of Some Rare Earth Metals and Their

Electron-Emission Properties

PERIODICAL:

V sb.: Bor. Tr. Konferentsii po khimii bora i yego soyedineniy. Moscow,

Goskhimizdat, 1958, pp 112 - 119

ABSTRACT:

The hexaborides of La, Cr, Pr, Nd and cerium-mixmetal can be obtained by the combined reduction of a mixture of the oxide of the corresponding

rare earth element and boron by carbon by means of thermal treatment under a certain condition (by stages). The emission constants of La and Ce hexaborides coincide well with the literature data; the constants of cerium-mixmetal boride deviate from them, which can be explained by the difference in the composition of the cerium-mixmetal samples. The coefficients of the secondary emission of all hexaborides are less than unity,

1.e., these hexaborides can be used for anti-dynatronic coatings, especially

Card 1/2

the hexaborides of Nd and Pr, which have also a low thermo-ionic emission

68953 SOV/81-60-2-4305

The Synthesis of Hexaborides of Some Rare Earth Metals and Their Electron-Emission Properties

activity. La hexaboride, due to the high thermo-ionic emission, can be used for the manufacture of cathodes for powerful superhigh-frequency devices. The low coefficient of secondary emission makes it impossible, however, to employ it for magnetronic cathodes. The radiation coefficients of all hexaborides are within the range 0.65 - 0.70. The hexaborides react with the underlaying material, forming Ta boride.

From the authors! summary

Card 2/2

80782

\$/137/60/000/01/01/009

Translation from: Referativnyy zhurnal, Metallurgiya, 1960, No 1, p 91, # 621

15,2226 AUTHORS:

Ormont, B.F., Epel'baum, V.A., Shafran, I.G.

TITLE:

Investigation, of the Boron-Carbon-Silicon System and Preparation

of Borundum V

PERIODICAL:

V sb.: Bor. Tr. Konferentsii po khimii bora i yego soyedineniy,

Moscow, Goskhimizdat, 1958, pp 177 - 181

To find ways of economizing the valuable B-raw material in the production of abrasive materials on B₄C base, the authors investigated the possibility of obtaining preparations containing B - C - Si, which are generally named "borundum". Preparations were studied which corresponded to the silicon vertex of the ternary structural diagram as well as preparations with a low (2 - 3%) Si content in B carbide. The preparations were produced in Tamman furnaces. B₂0₃ was obtained from boric acid, Si0₂ from ground white quartz and C from carbon black. The preparation corresponding to the Si2BC2 formula

Card 1/2

80782 S/137/60/000/01/01/009

Investigation of the Boron-Carbon-Silicon System and Preparation of Borundum

requires for its production an amount of B_2O_3 which is 6 times less than that necessary for B_hC_1 its efficiency is 80% of that of B_hC . The polishing efficiency of the "borundum"-type preparations exceeds that of carborundum by a factor of 5.

A.P.

Card 2/2

80777 s/137/60/000/03/05/013

Translation from: Referativnyy zhurnal, Metallurgiya, 1960, No 3, p 105, # 5312

AUTHORS:

Ormont, B.F., Epel'baum, V.A., Shafran, I.G.

15.2220 TITLE: Experience in the Industrial Borundum Production and Testing

of Its Properties

PERIODICAL:

V sb.: Bor. Tr. Konferentsii po khimii bora i yego soyedineniy,

Moscow, Goskhimizdat, 1958, pp 182 - 188

TEXT: For the purpose of raising the abrasive properties of carborundum, the authors carried out experimental smelts with admixture of B in the form of B_2O_3 (up to 8% and more) at temperatures slightly exceeding conventional temperatures. The smelts were prepared under industrial conditions in Acheson furnaces. It is shown that the product obtained - namely borundum - is very well fit for polishing and is 10 times cheaper than $B_{\rm h}C$. If small amounts of B are added the pressure of carborundum vapor changes noticeably, whereas the

Card 1/2

80777

\$/137/60/000/03/05/013

Experience in the Industrial Borundum Production and Testing of Its Properties

dimensions of the crystal lattice remain unchanged. The physical ground for the raised fitness to polishing of borundum in comparison to carborundum was as yet not found; however, data obtained do not confirm the hypothesis on the penetration of B atoms into the interstices of carborundum lattice.

X

A.P.

Card 2/2

SOV/78-3-11-19/23

AUTHORS:

Epel'baum, V. A., Sevast'yanov, N. G., Gurevich, M. A.,

Ormont, B. F., Zhdanov, G. S.

TITLE:

IL. On the Phases Formed in the System Chromium-Boron (II. O

fazakh, obrazuyushchikhsya v sisteme khrom-bor)

PERIODICAL:

Zhurnal neorganicheskoy khimii, 1958, Vol 3, Nr 11, pp 2545-2552

(USSR)

ABSTRACT:

The compounds formed in the system chromium-boron are investigated. The investigations were carried out by means of chemical, radiographic, and metallographic methods in the region of the phase diagram of the system chromium-boron and in the range CrB_{0.35}-CrB₃. Purest boron (99,6%) produced by the thermal dissociation of diboranes served as initial components for the production of the chromium-boron phases. The results of the

chemical and radiographic analyses of the samples were obtained by heating at 1150°C in vacuum and then at 1300°C in an argon atmosphere for 36 hours. The results are given in table 2. It was found that the Y-phase occurs with a rhombic lattice in the

Card 1/3

sample with a boron content of CrB_{0,35}-CrB_{0,58}. In the samples

CIA-RDP86-00513R000412120 APPROVED FOR RELEASE: Thursday, July 27, 2000

SOV/78-3-11-19/23

II. On the Phases Formed in the System Chromium-Boron

with a boron content of CrBO,41-CrBO,51 only the J-phase exists. In the samples with a boron content of CrBO,55-CrB1,05 the δ-phase (Cr₅B₃-phase) is formed. In the samples with a boron content of $CrB_{0,59}^{-CrB}_{0,63}$ only the δ -phase is formed. In the samples with a boron content of CrB_{0,68}-CrB_{1,50} the £-phase occurs (CrB with rhombic lattice). In the samples of the composition CrB 0,96 other phases were found besides the E-phase. In the sample with a boron content of CrB1,20 CrB 1.90 a 7-phase with rhombic lattice is formed. In the sample of the composition CrB_{1,50}-CrB_{1,65} no other phases were found to exist besides the ζ -phase. In the samples with CrB 1,70 and CrB_{1.90} only the \(\eta\)-phase is formed. There are 2 figures, 5 tables, and 27 references, 1 of which is

Card 2/3

5(4), 18(7)

AUTHORS: Epel'baum, V. A., Gurevich, M. A.

SOV/76-32-10-8/39

TITLE:

Investigation of the Phase Diagram of the System Zirconium - Boron (K issledovaniyu fazovoy diagrammy sistemy tsirkoniy-bor) II. On the Formation of the Phase as Dependent Upon the

Composition of ZrB, (II. Ob obrazovanii fazy, otvechayushchey

sostavu ZrB2)

PERIODICAL:

Zhurnal fizicheskoy khimii, 1958, Vol 32, Nr 10, pp 2274-2281

(USSR)

ABSTRACT:

Among the papers published in this field those by the following authors are mentioned and commented on: Tucker and Moody (Tuker, Mudi) (Ref 1), Wedekind (Vedekind) (Ref 2), Andrieux (Endro) (Ref 3), Moiers (Moyyers) (Ref 4), McKenna (Mak-Kenna) (Ref 5), Norton, Blumental and Sindeband (Blyumental', Zindeband) (Ref 6), Kiessling (Kissling) (Refs 7,8), Brewer, Sawzer,

(Ref 6), Kiessling (Kissling) (Refs 7,8), Brewer, Sawzer, Templeton and Dauben (Bryuyer, Sauzer, Templton and Daubin) (Ref 9), Kieffer, Benesovsky and Honak (Kiffer, Benesovski and Khonak) (Ref 10), G. A. Meyerson and G. V. Samsonov (Ref 11), as well as Post and Glaser (Glazer) (Refs 12-14). A zirconium of 99,6% (Zr 99,6%, Fe 0,07%, Ca 17%,001 0,001%) and boron

Card 1/3

SOV/76-32-10-8/39 Investigation of the Phase Diagram of the System Zirconium - Boron. II. On the Formation of the Phase as Dependent Upon the Composition of ZrB₂

(purity 99,3-99,5%) were used in the present experiments. The optimal sintering conditions are at 1900-2100°C and for a duration of 2-3 hours. The syntheses were carried out in a tungsten heater and in an argon atmosphere purified from humidity and oxygen, or in vacuum (10-3 mm Hg). The temperature measurements were carried out by means of the optical pyrometer "Ribo". M. I. Starostina and I. A. Pryanishnikova took part in the analyses. The composition of the α -phase is ZrB_{0.02} to ZrB_{2.68}. This phase represents a solid solution of boron up to 2 atom%) in hexagonal zirconium with the lattice triodes becoming greater. The ZrB phase also has a hexagonal ZrB0,02-0,03. The ZrB2 lattice and is already formed at phase is present from the composition ZrB1.7 to ZrB2.68 without any visible impurities of other boride phases of zirconium. The lattice periods of the phase ZrB, remain constant within the range of experimental error (Ref 15) (+ 0,001 kX) and amount to $a = 3,162_5 \pm 0,0003 \text{ kX}, c = 3,522_5 \pm 0,0003 \text{ kX}, c/a = 1,113_8.$

Card 2/3

Investigation of the Phase Diagram of the System Zirconium - Boron, II. On the Formation of the Phase as Dependent Upon the Composition of ZrB₂

Investigations carried out by an elutriation of ZrB_{2,68} in methylene iodide did not make possible a clear determination of whether this phase had a constant or variable composition. Data are given in tables and radiograms. Finally, the authors thank Professor B. F. Ormont. There are 1 figure, 3 tables, and 16 references, 3 of which are Soviet.

ASSOCIATION:

Fiziko-khimicheskiy institut im. L. Ya. Karpova Moskva (Moscow, Physical-Chemical Institute imeni L. Ya. Karpov)

Card 3/3

18(6) 807/78-4-6-31/44 AUTHORS: Epel'baum, V. A., Gurevich, M. A., Ormont, B. F. On the Nature of the $\alpha-$ and $\beta-$ phase Which Are Formed in the TITLE: System Boron-carbon (O prirode α- i β-faz, obrazuyushchikhsya v sisteme bor-uglerod) Zhurnal neorganicheskoy khimii, 1959, Vol 4, Nr 6, PERIODICAL: pp 1398 - 1403 (USSR) ABSTRACT: The conditions for the formation of the α - and β -phase in the system boron-carbon were investigated. Boron with a purity of 99.6% produced by the thermal dissociation of borane and carbon of a purity of 99.8% were used as initial products. The alloy was produced either in an argon atmosphere or in vacuum 10-3 torr at 1900 - 22000, and in vacuum 10-3 torr at 1150°, then stored for 10 hours at this temperature, and then stored 21 hours in an argon atmosphere at 1350°. Then the alloy was again heated and stored for three hours at 2300°. Table 1 shows the results of the chemical- and X-ray phase analyses of several preparations which were produced from purest initial products and purest boron anhydride. The Card 1/2 X-ray phase analysis showed that beside the line of the initial

On the Nature of the $\alpha-$ and $\beta-$ phase Which Are Formed in SOV/78-4-6-31/44 the System Boron-carbon

boron nitride also intensive lines of the $\alpha-$ and $\beta-$ phase occur in the products. The $\alpha-$ phase is coarse-grained, the $\beta-$ phase fine-grained. The influence of the thermal treatment of the boron carbide- $B_4^{\ C}$ - samples on the ratio of the $\alpha-$ and $\beta-$ phase was investigated as well as the graphite phase (Table 2). The results showed that a change of the matio in the lines of the $\alpha-$ and $\beta-$ phase occurs in the case of the thermal treatment in a Tammann furnace after the hot press method and in the furnace TVV-2. A mutual transformation of the $\alpha-$ and $\beta-$ phase takes place in the temperature range 1900 - 22000. The lattices of the $\beta-$ phase were more accurately determined and the average value a= 3.161+0.004 kX was detected. There are 2 tables and 5 references, 3 of which are Soviet.

SUBMITTED: March 27, 1958

Card 2/2

SOV/78-4-8-28/43

5(2) AUTHORS: Epel baum, V. A., Gurevich, M. A., Starostina, M. I.

TITLE:

On the Solubility of Boron in Silicon (O rastvorimosti bora

v kremnii)

PERIODICAL:

Zhurnal neorganicheskoy khimii, 1959, Vol 4, Mr 8, pp 1881-1884

(USSR)

ABSTRACT:

After a survey on the publication data dealing with this subject (Refs 1-14) the importance of the system mentioned in the title is pointed out since according to the publication data (Refs 5-7) the cermets which are produced from silicon and boron under the action of very high temperatures, are now industrially used. They are characterized by high strength, chemical stability, heat resistance, semiconductor properties, etc. The authors investigated the solubility of boron in silicon and its effect on the structure of the silicon crystal lattice. The composition of the samples was varied between 99Si: 1B and 1Si: 6B. The samples were melted at 1350° or 2100-2200°C in argon atmosphere and analysed by X-ray methods (X-ray camera RKU-86 and RKU-114, copper radiation \(\lambda \text{CuK}_{\alpha_1} = 1.537396 kX\). The lattice period of silicon decreasing with

Card 1/2

SOV/78-4-8-28/43

On the Solubility of Boron in Silicon

increasing boron content is shown by table 1 and graphically represented by using the data by F. Horn (Ref 14) and H. Nowotny (Ref 15) in figure 1. The behaviour of the solution of boron in silicon corresponds to the solid substitution solution. The strong contraction of the silicon lattice under the influence of relatively small boron amounts could not be explained. There are 1 figure, 1 table, and 17 references, 5 of which are Soviet.

SUBMITTED:

April 26, 1958

Card 2/2

EPEL BAUM, V.A.; SEVAST'YANOV, H.G.; GUREVICH, M.A.; ZHDANOV, G.S.

Phases formed in the system chromium - boron in the region rich in boron. Zhur. strukt. khim. 1 no.1:64-65 My-Je 160.

(MIRA 13:8)

1. Hauchno-issledovatel'skiy fiziko-khimicheskiy institut imeni L. Ya. Karpova. (Chromium) (Boron)

15.2410

28875 s/180/61/000/004/013/020 E071/E180

AUTHORS:

Card 1/4

Meyerson, G.A., Dergunova, V.S., Epel'baum, V.A.,

and Gurevich, M.A. (Moscow)

TITLE:

An investigation of some hard alloys of the

Boron-Silicon-Carbon system

PERIODICAL: Akademiya nauk SSSR. Izvestiya. Otdeleniye tekhnicheskikh nauk. Metallurgiya i toplivo.

90-94 no. 4, 1961,

The above system has, as yet, been insufficiently For this reason the authors investigated three groups of TEXT: alloys of the following types: alloys close to the zone of solid solutions based on SiC, alloys based on B4C, and alloys of the central part of the ternary B-Si-C system. In the latter, the points were chosen so as to overlap the zones in which previous investigators assumed the possibility of the existence of a ternary compound of the type BxSiyCz. Specimens of the alloys were obtained by hot pressing powder mixtures of the elements at 2000-2100 °C (no details of the preparation are given).

CIA-RDP86-00513R000412120 APPROVED FOR RELEASE: Thursday, July 27, 2000

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An investigation of some hard alloys.... \$/180/61/000/004/013/020
E071/E180

analysis of the specimens indicated that the sum of admixtures Porosity did not exceed (Fe, Mg, Al, Ca) did not exceed 0.1%. 2-5%, and density was uniform throughout the whole volume of the A prolonged high temperature annealing (50-100° below specimens. pressing temperature) brought the alloys to the equilibrium state with an increase in the grain size, but did not cause any changes in the chemical composition, or any increase in the porosity. The specimens were submitted to metallographic and X-ray analysis and microhardness measurements. The following conclusions are 1) A phase exists in the B-Si-C system with a melting drawn: temperature above 2100 °C and a very high hardness (about 7000 kg/mm^2 and above), noticeably exceeding the microhardness of boron carbide (5000 kg/mm^2). 2) In specimens produced and treated in the described way, metallographic and X-ray analysis did not show the presence of any new phases in noticeable quantities, only solid solutions based on B4C, SiC and Si (the latter at an insufficient carbon content). The X-ray analysis indicated that the solubility of silicon (or siliconcarbide) is small in boron carbide (less than 2% if calculated on Si), but metallographic investigation suggested Card 2/4

28875

An investigation of some hard alloys. \$\frac{5}{180}\frac{61}{000}\frac{004}{013}\frac{020}{020}

the presence of an apparently single phase up to 10-12% silicon. This can be explained by the separation of submicroscopically dispersed SiC particles on cooling. The microhardness of such grains, based on B4C, is 7000 kg/mm² and, in some cases reaches 8000 kg/mm2. 3) Grains of solid solutions based on SiC have a microhardness of 5000-5200 kg/mm2 instead of the 3500 of pure SiC. 4) The hardness of B-Si-C alloys changed little up to a temperature of 700-800 °C. For alloys based on B4C, the hardness of 6000-7000 kg/mm² at 20° dropped to 3000 kg/mm² at 800-900 °C and, for alloys based on SiC, from 4000-5000 kg/mm² to 1500 kg/mm². During these measurements, the formation of cracks was observed around the indentation in a number of cases, indicating that the actual hardness values could be higher. The work was carried out in the Kafedra redkikh metallov i poroshkovoy metallurgii (Department of Rare Metals and Powder Metallurgy) of the Institut tsvetnykh metallov imeni M.I. Kalinina (Institute of Non-ferrous Metals imeni M.I. Kalinin), in cooperation with the Fiziko-Khimicheskiy institut imeni L. Ya. Karpova (Physico-Chemical Institute imeni L.Ya. Karpov). Card 3/4

28875

s/180/61/000/004/013/020 An investigation of some hard alloys ... E071/E180

There are 1 figure, 1 table and 4 references: 3 Soviet-bloc and 1 English. The English language reference reads as follows:

Ref.1: F. Ton. The quest for hard materials. Industrial and Engineering Chemistry. Industrial edition, 1938, 30, 232-242.

SUBMITTED: November 21, 1960

Card 4/4

B/192/61/002/001/006/006 B107/B218

AUTHORS:

Epel'baum, V. A., Sevast'yanov, N. G., Ormont, B. F., and

Gurevich, M. A.

TITLE:

A possible existence of volume-centered phases of boron carbide

and silicon oxycarbide

PERIODICAL:

Zhurnal strukturnoy khimii, v. 2, no. 1, 1961, 65

TEXT: It has been stated in Ref. 1 (V. A. Epel'baum, M. A. Gurevich, B. F. Ormont, Zh. neorg. khimii, 1, 2149 (1956)) that lines of a cubic, volume-centered phase occur in preparations of boron carbide, which conclusion was drawn from the reflections of the X-ray picture. This volume-centered phase was called beta phase; it has a period of identity of 3.16 kX. The composition of this phase was not determined. The intensity of the reflections was very high for all samples, for some even higher than that of the reflections of the alpha phase. This led to the assumption that the beta phase belongs to the boron carbon system. The presence of impurities could, however, hardly be excluded, though every attempt was made to remove them (treatment with hydrofluoric and other acids). The authors of Ref. 2 (V. A. Epel'baum, M. A.

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Gurevich, B. F. Ormont, Zh. neorg. khimii, 4, 1938, (1959)) found that lines of this volume-centered phase occur in preparations with strongly differing content of boron and carbon. Thus, it was not possible to establish the position of the phase in the phase diagram of boron-carbon. This fact led to doubts about the composition of the phase, and thus to further experiments (see below). The authors of Ref 2 had pointed out that spectrum analysis did not show any considerable content of impurities. In 1958, Samsonov had published papers (Ref. 3: G. V. Samsonov, Zh. fiz. khimii, 32, 2424 (1958); Ref. 4: G. V. Samsonov, Ukr. khim. zh., 24, no. 6, 659 (1958), in which he stated already in 1952/1953 he had detected this phase in boron carbide, together with Zhuravlev, and found it to be silicon oxycarbide. Despite Samsonov's statement, this fact needs a further proof, especially since silicon oxycarbide is of practical, and the detection of Samsonov and Zhuravlev is of theoretical importance. Hitherto, only cubic silicon carbide and silicon oxycarbide have been known, both only with face-centered cell of the sphalerite type. A system of lines in the X-ray picture, however, corresponds to this structure which completely differs from that of the cubic, volume-centered cell. Thus, Samsonov claims to have detected a new phase of silicon oxycarbide with cubic, volume-centered cell and a period of identity Card 2/4

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of 3.16 kX. The authors of the present paper point out that a cubic, volumecentered cell with a period of identity of 3.16 kX leads to certain crystallochemical difficulties, both with boron carbide and silicon oxycarbide. This difficulty lies in the fact that the interatomic distance $d = a\sqrt{3/2} = 2.85$ kX is larger than the sum of the radii of the individual atoms. In order to explain this fact, it would be necessary to assume the existence of structural centers into which atom impurities enter, or one must assume the existence of complex structural centers with a corresponding system of reflections. The authors therefore arrived at the following conclusion: The system of reflections corresponding to a cubic, volume-centered cell of boron carbide is parasitic; it is formed by the occurrence of an additional phase in the preparation. By their careful experiments and control, the authors found that this admixture is introduced by the tungsten wire which is used for filling the sample to be studied radiographically into the capillary. For the first moment, it was striking that thereby such quantities of impurities could enter into the preparation that their lines are more intense than that of the main mass (Ref. 1). If, however, the great difference of the scattering power of tungsten as compared to boron, silicon, and carbon is considered, then the above effect, which was also observed by Card 3/4

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the authors of Ref. 2, becomes probable. One may assume that the cubic, volume-centered phase of silicon oxycarbide, which was detected by Samsonov and Zhuravlev (Ref. 3) in 1952, has the same origin. [Abstracter's note: This is a full translation from the original.] There are 4 Soviet-bloo references.

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Serebryanskiy, V. T., and Epel'baum, V. A.

TITLE:

AUTHORS:

Phase diagram for the system aluminum-boron

PERIODICAL:

Zhurnal strukturnoy khimii, v. 2, no. 6, 1961,

748-750

TEXT: The authors give the results of the first stage of their study of the phase diagram for the system Al-B and the conditions of the synthesis of aluminum borides at temperatures of up to 1400°C. Of the scientists, who also investigated the borides of aluminum only E. J. Felfen succeeded in obtaining AlB, as the main

reaction product during the sintering of Al and B; moreover, there is no information in literature regarding the conditions of existence of other aluminum-boride phases. Mixed and compressed Al-B powder was first heated for 30 min. in a vacuum at about 400°C. Purified argon was then admitted, after which the reaction mass was

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Phase diagram for the system ...

heated at higher temperatures. In addition to the X-ray analysis of the pure phases and the determination of their pycnometric density, part of the borides was studied thermographically in an apparatus designed by N. A. Nedumov. The experimental results show that the formation of AlB₂ starts at 650°, and that the most suitable temperature for the process is 800°. AlB₂ begins to decompose at higher temperatures: after prolonged heating at 950° the reaction products consist of AlB₂, Al, and AlB₁₂. The data of the thermal analysis were also confirmed by further tests which disclosed the formation of AlB₂ and d-AlB₁₂ a tetragonal modification containing 82.8% B and 16.7% Al, at temperatures of \triangleleft 900° and \triangleright 1000° respectively. The pycnometric density of \triangleleft -AlB₁₂ was found to be 2.62 g./cm². In conclusion the authors state that the conditions of formation of AlB₁₂ and AlB₁₀ will be determined in a subsequent study.

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